

## Note

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### Mode of action of dextranase D<sub>2</sub> from *Pseudomonas* UQM 733 on oligosaccharides\*

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Well-characterised, pure dextranases offer great advantages to the study of the fine structure of dextrans. However, although there are many recent reports of dextranases induced in various fungi and bacteria, few workers have been concerned with the rigorous purification and subsequent characterisation of the mode of action of these enzymes<sup>1</sup>.

We have described<sup>2,3</sup> the isolation of two extracellular endo-dextranases (D<sub>1</sub> and D<sub>2</sub>) from a species of *Pseudomonas* (UQM 733). The mode of action of pure dextranase D<sub>1</sub> has been studied and a model for its catalytic centre proposed. We now describe similar work with dextranase D<sub>2</sub>, which differs from D<sub>1</sub> in a number of ways.

Purification<sup>2</sup> of D<sub>2</sub> by gel-permeation chromatography gave a product which showed two protein bands on gel electrophoresis. Only the major band had dextranase activity. It was not possible to remove the non-dextranase protein from D<sub>2</sub>, but no glycanase activity was detected, other than that against (1 → 6)- $\alpha$ -D-glucan, for a wide range of polysaccharide types. The preparation is also free from activity against sucrose, maltose, and cellobiose. Thus, D<sub>2</sub> appears to be suitable for use in structural studies on dextrans, and its specificity and action pattern against a series of oligosaccharides have been investigated.

The time course of degradation of isomaltosaccharides of d.p. 7–12 (IM<sub>7</sub>–IM<sub>12</sub>) is shown in Fig. 1, and the products are indicated in Table I. The rate of degradation and the percentage increase in reducing power (*i.e.*, scissions per molecule) increased with increase in d.p. 7 → 12, but a maximum was not attained. Although IM<sub>2</sub>–IM<sub>6</sub> showed no measurable increase in reducing power on incubation with D<sub>2</sub>, p.c. of the IM<sub>5</sub> and IM<sub>6</sub> digests indicated slight attack by the enzyme during 3 days. Each substrate was degraded slightly to IM<sub>2</sub>, IM<sub>3</sub>, and IM<sub>4</sub> with trace amounts of glucose, whereas IM<sub>6</sub> also gave small proportions of IM<sub>5</sub>. Little

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\*Studies on Dextranases: Part VI. For Part V, see Ref. 7.

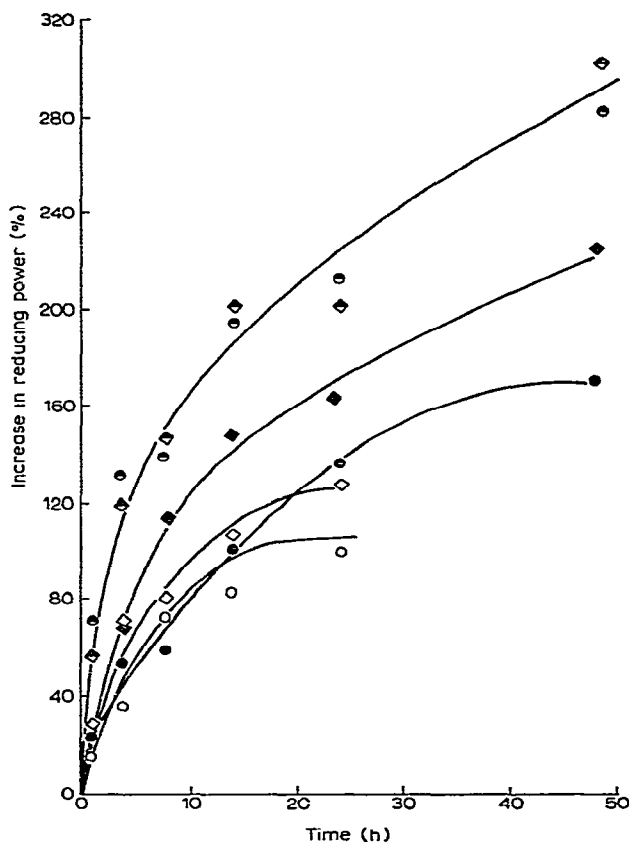


Fig. 1. Degradation of oligosaccharides by dextranase  $D_2$  at  $33^\circ$ : —○—  $IM_7$ , —●—  $IM_8$ , —◇—  $IM_9$ , —◆—  $IM_{10}$ , —◻—  $IM_{11}$ , —◼—  $IM_{12}$ ;  $3.8 \pm 0.3\text{mM}$ ;  $D_2$   $2.3 \times 10^{-2}$  unit.

or no glucose, isomaltose, or oligosaccharides of d.p. one or two less than the substrate were produced from the action of  $D_2$  against substrates of d.p.  $>7$  and thus, in contrast to  $D_1$ ,  $D_2$  did not hydrolyse glycosidic linkages close to either end of the substrate chain. The  $D_2$  enzyme showed a marked tendency to hydrolyse centrally located glycosidic linkages. This property was most obvious on degradation of  $IM_{10}$ ,  $IM_{11}$ , and  $IM_{12}$ , where no significant products of d.p.  $<3$  were evident. There was no evidence for any synthetic activity such as was observed with  $D_1$ .

Fig. 2 shows the yields of products formed on degradation of  $IM_{10}$  with  $D_2$ .  $IM_5$  was produced most rapidly ( $84 \times 10^{-3} \mu\text{mol.h}^{-1}$ ) and must derive from hydrolysis of the central glycosidic linkage.  $IM_4$  and  $IM_6$  were produced initially at the same rate (combined rate,  $65 \times 10^{-3} \mu\text{mol.h}^{-1}$ ), which was greater than that for  $IM_7$  and  $IM_3$  (combined rate,  $25 \times 10^{-3} \mu\text{mol.h}^{-1}$ ). Hydrolysis of the central glycosidic linkage was therefore the preferred mode of attack of  $D_2$  on  $IM_{10}$ , followed by hydrolysis of other linkages at a rate decreasing with distance from the central linkage.

TABLE I

PRODUCTS OF DEGRADATION OF ISOMALTOSE OLIGOSACCHARIDES<sup>a</sup> ( $3.8 \pm 0.3\text{mm}$ ) BY DEXTRANASE D<sub>2</sub> ( $2.3 \times 10^{-3}$  UNIT)

Products	IM <sub>2</sub>	IM <sub>3</sub>	IM <sub>4</sub>	IM <sub>5</sub>	IM <sub>6</sub>	IM <sub>7</sub>	IM <sub>8</sub>	IM <sub>9</sub>	IM <sub>10</sub>	IM <sub>11</sub>	IM <sub>12</sub>
D-Glucose	—	— <sup>c</sup>	—	—	—	—	—	—	—	—	—
IM <sub>2</sub>	4	4	—	—	—	—	—	—	—	—	—
IM <sub>3</sub>	—	—	—	—	—	—	—	—	—	—	—
IM <sub>4</sub>	—	4	—	—	—	—	—	—	—	—	—
IM <sub>5</sub>	—	—	4	—	—	—	—	—	—	—	—
IM <sub>6</sub>	—	—	—	4	—	—	—	—	—	—	—
IM <sub>7</sub>	—	—	—	—	—	—	—	—	—	—	—
IM <sub>8</sub>	—	—	—	—	—	—	—	—	—	—	—
IM <sub>9</sub>	—	—	—	—	—	—	—	—	—	—	—
IM <sub>10</sub>	—	—	—	—	—	—	—	—	—	—	—
IM <sub>11</sub>	—	—	—	—	—	—	—	—	—	—	—
IM <sub>12</sub>	—	—	—	—	—	—	—	—	—	—	—

<sup>a</sup>IM<sub>x</sub> connotes an isomaltosaccharide containing x glucose residues. <sup>b</sup>The figures represent a visual estimation of the amounts after p.c. The pair of numbers in each column refers to samples taken after 4 h and 2 days, respectively. <sup>c</sup>Samples taken after 3 days.

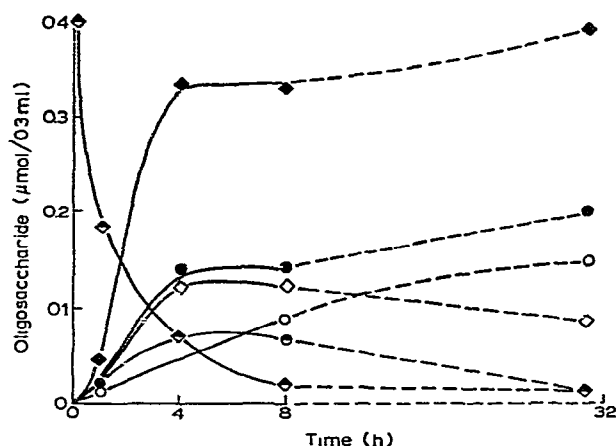


Fig. 2. Degradation of IM<sub>10</sub> (2 mM) by dextranase D<sub>2</sub> ( $13.7 \times 10^{-2}$  unit initially,  $6.5 \times 10^{-2}$  unit added after 24 h): —○— IM<sub>3</sub>, —●— IM<sub>4</sub>, —◆— IM<sub>5</sub>, —◇— IM<sub>6</sub>, —●— IM<sub>7</sub>, —◆— IM<sub>10</sub>.

TABLE II

DEGRADATION OF REDUCED, TRITIATED IM<sub>10</sub> (6mM) WITH DEXTRANASE D<sub>2</sub> ( $2.3 \times 10^{-2}$  UNIT)

Reduced, tritiated oligosaccharide	Counts/min (% total)	Frequency of production <sup>a</sup>
IM <sub>2</sub> -itol	1	0.01
IM <sub>3</sub> -itol	13	0.13
IM <sub>4</sub> -itol	23	0.23
IM <sub>5</sub> -itol	41	0.41
IM <sub>6</sub> -itol	15	0.15
IM <sub>7</sub> -itol	4	0.04
IM <sub>10</sub> -itol	3	

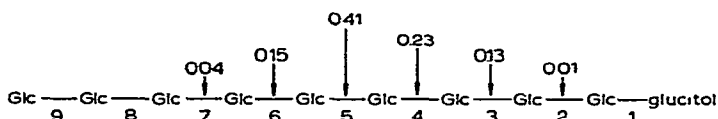
<sup>a</sup>Counts (%) in each oligosaccharide product/total counts (%) for oligosaccharide products.

In addition (*cf.* Table I), there was some secondary hydrolysis of the primary products from IM<sub>10</sub>, *e.g.*, IM<sub>7</sub> and IM<sub>6</sub>, as indicated by their decrease in concentration after prolonged incubation times.

It was not possible to determine whether or not the rates of bond scission in IM<sub>10</sub> were symmetrical on either side of the central bond because, for example, scission of bond 4 (see below) gives the same products as scission of bond 6. However, the fit of the oligosaccharide substrates into the enzyme binding-site is likely to be unidirectional. In order to resolve this point, IM<sub>10</sub> was reduced with sodium borotritide, thereby converting the reducing end-group into D-glucitol-1-*l*.

Table II shows the distribution of radioactive products obtained on degradation of reduced IM<sub>10</sub> by D<sub>2</sub>, and the relative frequency of production of each as calculated by the method of Robyt and French<sup>4</sup>. The calculation assumes that all of the radioactive products are derived from the primary scission of reduced IM<sub>10</sub>. This assumption is considered valid, because the largest, initial, tritiated product (reduced IM<sub>7</sub>) is degraded<sup>5</sup> at half the rate of IM<sub>7</sub>, which, in turn, is degraded slowly compared to IM<sub>10</sub>. Thus, the presence of an alditol group at the reducing end of isomaltosaccharides of d.p.  $\leq 7$  renders the molecules relatively stable to attack by D<sub>2</sub>.

The scission of reduced IM<sub>10</sub> by D<sub>2</sub> can be represented diagrammatically as follows:



Thus, bond 5 was hydrolysed most rapidly.

The alditol grouping in reduced IM<sub>10</sub> did not prevent formation of an active enzyme-substrate complex; in fact, more bond cleavage occurred on the side towards the alditol group. Thus, the fit of IM<sub>10</sub> and reduced IM<sub>10</sub> in the active site appears to be similar and there is a similar asymmetrical pattern of rates of bond scission. A more complex asymmetry of distribution of hydrolysis rates along the oligosaccharide chain has been observed by Walker and Dewar<sup>6</sup> in the degradation of substrates, labelled at the non-reducing end, by *Penicillium lilacinum* dextranase.

Whereas dextranase D<sub>1</sub> hydrolyses smaller isomaltosaccharides ( $\geq$  IM<sub>5</sub>) and reaches a maximum rate of attack with IM<sub>8</sub>, D<sub>2</sub> shows no maximum rate of attack with oligosaccharides of d.p. 7–12, with IM<sub>7</sub> being the smallest isomaltosaccharide to be hydrolysed at a significant rate. Both dextranases show a preference for hydrolysis of glycosidic linkages towards the reducing end of the susceptible oligosaccharides that are degraded at less than the maximum rate. This pattern reflects the nature of the active site which, for D<sub>1</sub>, has been postulated<sup>3</sup> to consist of six glucose-binding subsites arranged asymmetrically about the catalytic site. With oligosaccharides of d.p.  $>6$ , the subsites can be filled in more than one way and hydrolysis is more random, with the centrally located linkages being hydrolysed most frequently.

It is not possible to postulate an exact model for the active centre of D<sub>2</sub>, as the largest oligosaccharide available, IM<sub>12</sub>, did not indicate that a maximal rate of hydrolysis had been attained. However, the available results suggest that D<sub>2</sub> has a much larger active centre than does D<sub>1</sub> and that ten or more glucose-binding subsites could be involved. The results of degradation of reduced IM<sub>10</sub> suggest that the catalytic site is asymmetrically placed within the binding subsites, probably towards the reducing end. On this basis, and assuming that the active site will not accommodate a branch point in a dextran molecule, it is unlikely that D<sub>2</sub> will hydrolyse dextran molecules at a significant rate, except in regions containing an unbranched sequence of at least seven (1  $\rightarrow$  6)-linked  $\alpha$ -D-glucosyl residues. This conclusion

has been used to derive information on the distribution of branch points in B-512 dextran<sup>7</sup>.

#### EXPERIMENTAL

General methods and the preparation of substrates have been described earlier<sup>2,3</sup>.

*Enzymic hydrolysis of isomaltose oligosaccharides.* — A solution (0.5 ml) of the oligosaccharide (3.8mM, d.p. 2–12) and dextranase D<sub>2</sub> ( $2.3 \times 10^{-2}$  unit) in 54mM sodium citrate buffer (pH 5.5) was kept at 33°, and reducing power was monitored by the Nelson–Somogyi method<sup>8</sup> and paper chromatography (p.c.).

For the quantitative determinations, a solution (2.6 ml) containing 54mM citrate buffer (pH 5.5), D<sub>2</sub> ( $13.7 \times 10^{-2}$  unit), and IM<sub>10</sub> (1.35mM) was kept at 33°, and 300- $\mu$ l samples were removed after 1, 4, and 8 h. After 24 h, more enzyme ( $6.5 \times 10^{-2}$  unit) was added, and the final sample was removed after a further 8 h. The relative amounts of the different oligosaccharides in each sample were determined as described previously.

*Enzymic hydrolysis of tritiated, reduced isomaltodecaose.* — A solution (0.5 ml) containing 54mM citrate buffer (pH 5.5), 6.0mM tritiated, reduced isomaltodecaose (9,000 counts.min<sup>-1</sup>. $\mu$ mol<sup>-1</sup>), and D<sub>2</sub> ( $2.3 \times 10^{-2}$  unit) was kept at 33°, and 100- $\mu$ l samples were removed after 1 and 24 h. The relative amounts of radioactive products were determined as described previously.

*Activity of dextranase D<sub>2</sub> against other poly- and di-saccharides.* — D<sub>2</sub> ( $2.1 \times 10^{-2}$  unit) was incubated with 1% solutions of potato starch, amylose, amylopectin, glycogen, pullulan, carboxymethylcellulose, laminaran, nigeran, hemicellulose B from sugar cane, sucrose, cellobiose, or maltose in 54mM citrate buffer (pH 5.5) at 40° for 24 h. The reducing power of the solutions was then measured, and compared with the control solutions lacking enzyme.

#### REFERENCES

- 1 R. L. SIDEBOTHAM, *Adv. Carbohydr. Chem. Biochem.*, 30 (1974) 371–444.
- 2 G. N. RICHARDS AND M. STREAMER, *Carbohydr. Res.*, 25 (1972) 323–332.
- 3 G. N. RICHARDS AND M. STREAMER, *Carbohydr. Res.*, 32 (1974) 251–260.
- 4 J. F. ROBYT AND D. FRENCH, *J. Biol. Chem.*, 245 (1970) 3917–3927.
- 5 M. STREAMER, Ph.D. Thesis, James Cook University of North Queensland, 1973.
- 6 G. J. WALKER AND M. D. DEWAR, *Carbohydr. Res.*, 39 (1975) 303–316.
- 7 M. T. COVACEVICH AND G. N. RICHARDS, *Carbohydr. Res.*, 54 (1977) 311–315.
- 8 N. NELSON, *J. Biol. Chem.*, 153 (1944) 375–380.